Zuschriften

Ring-Opening Catalysts

Cationic Planar Chiral Palladium P,S Complexes as Highly Efficient Catalysts in the Enantioselective Ring Opening of Oxa- and Azabicyclic Alkenes**

Silvia Cabrera, Ramón Gómez Arrayás, and Juan C. Carretero*

A huge effort has been devoted to the development of highly efficient asymmetric transition-metal-catalyzed reactions using chiral ligands based on P,P, P,N or N,N coordination modes, [1] some of which have reached the status of privileged ligands. [2] In sharp contrast, thioether-based chiral ligands have received much less attention, despite a key structural

[*] S. Cabrera, Dr. R. Gómez Arrayás, Prof. Dr. J. C. Carretero Departamento de Química Orgánica Facultad de Ciencias, Universidad Autónoma de Madrid Cantoblanco 28049 Madrid (Spain) Fax: (+34) 914-973-966 E-mail: juancarlos.carretero@uam.es

[**] This work was supported by the MCyT (project BQU2000-0226 and BQU2003-0508). S.C. thanks the UAM for a predoctoral fellowship. R.G.A. thanks the MCyT for a Ramón y Cajal contract. We acknowledge the skilled assistance of Prof. Jaime Rodríguez González in solving the X-ray crystal structure of the cationic palladium complex. Prof. José L. García-Ruano is gratefully acknowledged for unrestricted access to the HPLC equipment of his group. We thank Johnson Matthey PLC for a generous loan of PdCl₂.

Supporting information for this article is available on the WWW under http://www.angewandte.org or from the author.

feature that makes them very attractive in asymmetric catalysis: the sulfur atom becomes stereogenic upon coordination to the metal, thus imposing a unique asymmetric environment next to the reactive metal center. In this context, we recently reported that the planar chiral Fesulphos ligands 1 (1-phosphanyl-2-sulfenylferrocenes) behave as very efficient P,S ligands in Pd-catalyzed allylic substitutions and in Cu-catalyzed additions to imines.^[4]

Among recently described new Pd-catalyzed enantioselective reactions, the ring opening of meso oxabicyclic alkenes with dialkyl zinc reagents in the presence of chiral P,P and P,N ligands reported by Lautens et al. constitutes a synthetically outstanding C-C bond forming desymmetrization reaction.^[5] According to mechanistic evidence^[5b] this alkylative ring-opening reaction proceeds by coordination and subsequent enantioselective carbopalladation of the alkene, with a cationic alkyl palladium intermediate [L₂PdR]⁺ formed in situ under the reaction conditions employed. [6] Inspired by this work, we envisaged that the cationic methylpalladium complexes of Fesulphos ligands [(1)PdMe]+ could act as very active catalysts in this type of reaction. Herein, we report that this robust and readily available catalyst system combines low catalyst loading, high enantioselectivity, easy fine-tuning, mild reaction conditions, and broad structural scope.

Treatment of ligands 1 with [PdCl₂(CH₃CN)₂]^[7] afforded in very high yield the corresponding complexes [(1)PdCl₂] as single epimers at the sulfur atom. [4b] Interestingly, the transmetalation reaction of these complexes with Me₂Zn was completely stereoselective, affording a single complex [(1)Pd(Cl)(Me)] in 70–95% yield (Scheme 1). The cis arrangement of the methyl group and the phosphane moiety was established by analysis of their ¹H, ¹³C, and ³¹P NMR spectra; the low coupling constants ${}^2J_{\rm P,C(Me)}$ (< 1.0 Hz) and ³J_{PH(Me)} (2.4 Hz) were of great diagnostic value. [8] This stereochemical assignment was unequivocally confirmed by X-ray crystal diffraction analysis^[9] of [(1a)Pd(Cl)(Me)] (Figure 1). Additional relevant structural information includes the preservation of the anti arrangement of the tert-butyl group with respect to the iron atom ($R_{\rm p}R_{\rm s}$ configuration), and the elongation of the Pd-S bond (2.444 Å) compared to that of the starting dichloro complex (2.314 Å)[4b] as result of the strong trans effect of the highly σ-donating methyl group. The removal of the chloride ion in [(1a)Pd(Cl)(Me)] by treatment with AgPF₆ (CH₂Cl₂, room temperature) or NaB(Ar^F)₄ $[Ar^F = 3,5$ -bis(trifluoromethyl)phenyl]^[10] in the presence of benzonitrile afforded a single cationic benzonitrile-coordinated palladium species [(1a)Pd(Me)(PhCN)]+X⁻. The crystal structure of this complex^[11] (Figure 1) is very similar to that of its precursor [1a·Pd(Cl)(Me)] although, as expected, the Pd-Me bond is even shorter (2.011 versus 2.048 Å) and the Pd-S bond is longer (2.461 versus 2.444 Å) as a result of the enhanced σ donation of the methyl group to the cationic Pd atom.

Table 1 summarizes the most significant results obtained with this set of potential P,S catalysts in the model reaction of 7-oxabenzonorbornadiene with Me₂Zn.

As the starting point we used similar experimental conditions to those previously described: 5 mol % of catalyst in toluene or 1,2-dichloroethane (DCE) at room temperature

Scheme 1. Synthesis of Fesulphos palladium complexes. a) Me_2Zn (1.5 equiv), CH_2Cl_2 , RT; b) $AgPF_6$ or $NaBAF_4$, PhCN, CH_2Cl_2 , RT. Cy = cyclohexyl.

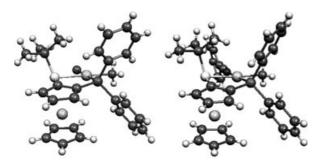


Figure 1. X-Ray crystal structures of [(1 a)Pd(Cl)(Me)] and $[(1 \text{ a})\text{Pd}(\text{Me})(\text{PhCN})]\text{B}(\text{Ar}^{\text{F}})_4$ (the $\text{B}(\text{Ar}^{\text{F}})_4^-$ anion has been omitted for clarity).

(RT).^[5] Catalyst [(1a)PdCl₂] led to 80% conversion in 24 hours (entry 1) to give the *cis* alcohol (*R*,*R*)-2a^[12] with a remarkable 81% *ee*. A similar result was achieved using the methyl complex [(1a)Pd(Cl)(Me)] (entry 2), which supports the proposal that both complexes evolve to the same active catalyst under the reaction conditions used. A significant increase in the reaction rate occurred when a catalytic amount of AgPF₆ was added to dissociate the chloride ligand (entry 3). However, the greatest acceleration effect was

produced by adding NaB(Ar^F)₄ (entry 4). A complete conversion was observed within 10 minutes with this noncoordinating counterion. This outstanding reactivity allowed a dramatic decrease of the catalyst loading: 0.5 mol % of catalyst drove the reaction to completion in 30 minutes (entry 5). As expected, a similar result was obtained using 0.5 mol % of the isolated air-stable cationic complex (entry 6) instead of catalyst generated in situ.

To fine-tune the reactivity and enantioselectivity of the process, we next studied the effect of the substitution at the phosphorus atom (complexes of ligands **1b–e**). With the exception of the complex containing the bulky naphthylphosphane ligand **1d** (entry 9), the reactions were complete in 10–30 minutes, with the electronically rich phosphane **1e** providing the best enantioselectivity (entry 10). The combination [(**1e**)Pd(Cl)(Me)]+NaB(Ar^F)₄ proved to be so effective that the reaction could be performed with severely decreased catalyst loading and temperature, which resulted in an enhancement of the enantioselectivity. Thus, 0.2 mol% of catalyst was sufficient to reach quantitative conversion within 5 h at -25 °C, which afforded alcohol **2a** in 88 % yield with 97 % *ee* (entry 11).

With these optimized P,S catalysts in hand, we turned our attention toward the substrate generality of the process. The use of 0.5 mol % $[(1e)Pd(Cl)(Me)]^{[13]}$ in combination with NaB(Ar^F)₄ (0.5 mol %) as a halogen scavenger catalyzed the enantioselective opening of a variety of substituted *meso* oxabenzonorbornadienes with both Me₂Zn and Et₂Zn in DCE^[14] at room temperature. As shown in Scheme 2, complete conversions within 10–30 minutes, good chemical yields (61–98%), and excellent enantioselectivities (94– >99% *ee*, HPLC) were obtained for all products 2–5.

Table 1: Ring opening of 7-oxabenzonorbornadiene with Me₂Zn catalyzed by palladium complexes of ligands 1.

| Entry | Catalyst (x mol%) | Additive (x mol%) | t (min) | Yield [%] ^[a] | ee [%] ^[b] |
|-------------------|--|---------------------------------------|---------|--------------------------|-----------------------|
| 1 | [(1 a) PdCl ₂] (5) | _ | 1440 | 80 ^[c] | 81 |
| 2 | [(1 a) Pd(Cl) (Me)] (5) | _ | 2880 | 75 ^[c] | 85 |
| 3 | [(1 a) Pd(Cl)(Me)] (5) | $AgPF_6$ (5) | 420 | 83 | 83 |
| 4 | [(1 a) Pd(Cl) (Me)] (5) | NaBAr ^F ₄ (5) | 10 | 86 | 78 |
| 5 | [(1 a) Pd(Cl)(Me)] (0.5) | NaBAr ^F ₄ (0.5) | 30 | 78 | 78 |
| 6 | $[(1 \text{ a}) Pd(Me) (PhCN)]^+(BAr^{F_4})^- (0.5)$ | _ | 30 | 82 | 72 |
| 7 | [(1 b)Pd(Cl)(Me)] (5) | NaBAr ^F ₄ (5) | 15 | 71 | 69 |
| 8 | [(1 c)Pd(Cl)(Me)] (5) | NaBAr ^F ₄ (5) | 30 | 74 | 64 |
| 9 | [(1 d)Pd(Cl)(Me)] (5) | NaBAr ^F ₄ (5) | 360 | 80 | 46 |
| 10 | [(1 e) Pd(Cl) (Me)] (5) | NaBAr ^F ₄ (5) | 10 | 77 | 83 |
| 11 ^[d] | [(1 e) Pd (Cl) (Me)] (0.2) | NaBAr ^F ₄ (0.2) | 300 | 88 | 97 |

[a] In pure product. [b] HPLC (Chiralpak AD column). [c] Conversion yield. [d] In 1,2-dichloroethane at -25 °C.

Zuschriften

2a: R = Me, 88%, 97% ee^(a) **3a**: R = Me, 71%, >99% ee **4a**: R = Me, 95%, 95% ee **5a**: R = Me, 98%, 97% ee **2b**: R = Et, 81%, 95% ee **3b**: R = Et, 61%, 94% ee **4b**: R = Et, 79%, 96% ee **5b**: R = Et, 85%, 94% ee

Scheme 2. Alcohols **2–5** resulting from the reaction of substituted oxabenzonorbornadienes with Me_2Zn and Et_2Zn catalyzed with $[(1e)PdMe]^+$. [a] The reaction was carried out at -25 °C.

Finally, to test the efficiency of these highly active catalysts toward much less reactive bicyclic substrates, we studied the opening reaction of nonaromatic [2.2.1]oxabicyclic alkenes and azabenzonorbornadienes, which usually require harsher reaction conditions. [Sc] Notably, [(1a)PdMe]⁺ and [(1e)PdMe]⁺ (1 mol%) induced the ring opening of the [2.2.1]oxabicyclic alkene with Me₂Zn within three hours at room temperature, which led to cyclohexenol 6a^[12] in 96–97% *ee* (Scheme 3).

Scheme 3. Ring opening of less reactive *meso* substrates. TBDPS = *tert*-butyldiphenylsilyl, Ts = toluene-4-sulfonyl. [a] The reaction was carried out with 5 mol% of catalyst.

The most reactive catalyst for the opening of the azabenzonorbornadiene derivative was the complex $[(1\mathbf{a})\mathrm{Pd}(\mathrm{Me})(\mathrm{PhCN})]^+(\mathrm{PF}_6)^-$. In the reaction with $\mathrm{Me}_2\mathrm{Zn}$ the ring-opened product (R,R)- $7\mathbf{a}$ was obtained within 30 minutes (73 % yield) with virtually complete enantiocontrol (>99 % ee) by using 5 mol % of catalyst. The use of 1 mol % of catalyst resulted in a much slower reaction (60 % conversion after 48 h). The reaction with $\mathrm{Et}_2\mathrm{Zn}$ was faster, but surprisingly much less enantioselective, and gave $7\mathbf{b}$ in 33 % ee.

According to the X-ray structures shown in Figure 1, the high enantioselectivity displayed by these P,S-palladium species could be explained by assuming a combination of electronic and steric factors: a) coordination of the bicyclic alkene *trans* to the phosphorus atom on the palladium center; and b) coordination of the alkene from its less hindered *exo* face, with orientation of the oxygen (or nitrogen) bridge to the opposite side of the very bulky *tert*-butyl group. Thus, the synergetic effect derived from the strong electronic *trans*

effect of the phosphorus moiety and the great steric control exerted by the close, bulky stereogenic sulfur substituent would be responsible for the high asymmetric induction in the key carbopalladation step (Scheme 4).

In summary, the cationic methylpalladium species derived from the readily available planar chiral P,S ligands 1 display an excellent profile as catalysts for the enantioselective alkylative ring

MeZn
$$(R, R)$$
 (R, R) $(R, R$

Scheme 4. Mechanistic proposal.

opening of oxa- and azabicyclic alkenes with dialkyl zinc reagents. The structure of these catalysts suggests that the high asymmetric induction relies on the strong *trans* effect of the phosphane moiety that acts in combination with the sterically demanding environment imposed by the stereogenic sulfur atom directly bonded to the palladium atom.

Received: March 22, 2004 [Z460087]

Keywords: alkenes \cdot asymmetric catalysis \cdot ferrocenyl ligands \cdot palladium \cdot S ligands

- a) Comprehensive Asymmetric Catalysis (Eds.: E. N. Jacobsen, A. Pfaltz, H. Yamamoto), Springer, Berlin, 1999;
 b) Catalytic Asymmetric Synthesis (Ed.: I. Ojima), Wiley, New York, 2000.
- [2] T. P. Yoon, E. N. Jacobsen, *Science* **2003**, *299*, 1691–1693.
- [3] For leading recent references on P,S ligands in enantioselective catalysis, see: sulfenyl phosphinites: a) D. A. Evans, F. E. Michael, J. S. Tedrow, K. R. Campos, J. Am. Chem. Soc. 2003, 125, 3534–3543; b) T. Kanayama, K. Yoshida, H. Miyabe, Y. Takemoto, Angew. Chem. 2003, 115, 2100–2102; Angew. Chem. Int. Ed. 2003, 42, 2054–2056; c) D. A. Evans, K. R. Campos, J. S. Tedrow, F. E. Michael, M. R. Gagne, J. Am. Chem. Soc. 2000, 122, 7905–7920; sulfenyl phosphines: d) T. Tu, Y.-G. Zhou, X.-L. Hou, L.-X. Dai, X.-C. Dong, Y.-H. Yu, J. Sun, Organometallics 2003, 22, 1255–1265; e) H. Nakano, Y. Suzuki, C. Kabuto, R. Jujita, H. Hongo, J. Org. Chem. 2002, 67, 5011–5014; f) D. Enders, R. Peters, R. Lochtman, G. Raabe, J. Runsik, J. W. Bats, Eur. J. Org. Chem. 2000, 3399–3426; g) X. Verdaguer, A. Moyano, M. A. Pericàs, A. Riera, M. A. Maestro, J. Mahía, J. Am. Chem. Soc. 2000, 122, 10242–10243.
- [4] a) O. García Mancheño, R. Gómez Arrayás, J. C. Carretero, J. Am. Chem. Soc. 2004, 126, 456–457; b) O. García Mancheño, J. Priego, S. Cabrera, R. Gómez Arrayás, T. Llamas, J. C. Carre-

- tero, *J. Org. Chem.* **2003**, *68*, 3679 3686; c) J. Priego, O. García Mancheño, S. Cabrera, R. Gómez Arrayás, T. Llamas, J. C. Carretero, *Chem. Commun.* **2002**, 2512 2513.
- [5] a) M. Lautens, S. Hiebert, J. Am. Chem. Soc. 2004, 126, 1437–1447; b) M. Lautens, S. Hiebert, J.-L. Renaud, J. Am. Chem. Soc. 2001, 123, 6834–6839; c) M. Lautens, S. Hiebert, J.-L. Renaud, Org. Lett. 2000, 2, 1971–1973; d) M. Lautens, J.-L. Renaud, S. Hiebert, J. Am. Chem. Soc. 2000, 122, 1804–1805. For a review on enantioselective metal-catalyzed opening reactions of oxabicyclic alkenes, see: M. Lautens, K. Fagnou, S. Hiebert, Acc. Chem. Res. 2003, 36, 48–58.
- [6] According to the participation of cationic palladium species, it has been recently reported that the cationic μ -hydroxo complex $[\{Pd((R)\text{-binap})-(\mu\text{-OH})\}_2]^{2+}(OTf^-)_2$ is much more effective than the currently used dichloro complex, see reference [5a] (binap = 2,2'-bis(diphenylphosphanyl)-1,1'-binaphthyl, Tf = trifluoromethanesulfonyl).
- [7] For the use of the parent ligands 1 in the Pd-catalyzed desymmetrization of oxabenzonorbornadiene, see reference [4c].
- [8] For achiral cationic methyl P,S-palladium complexes, see: a) O. Daugulis, M. Brookhart, P. S. White, *Organometallics* 2002, 21, 5935–5943; b) G. P. Suranna, P. Mastrorilli, C. F. Nobile, W. Keim, *Inorg. Chim. Acta* 2000, 305, 151–156.
- [9] Crystal structure for $C_{27}H_{30}$ CIFePPdS [(1a)Pd(Cl)(Me)]: crystal size $0.35\times0.34\times0.32$ mm, hexagonal, space group $P6_1$, a=18.3144(2), b=18.3144(2), c=15.7573(2) Å, $\gamma=120^\circ$, V=4577.18(9) ų, Z=6, $\rho_{\text{calcd}}=1.339$ Mg m³, $\mu=10.583$ mm¹, $2\theta_{\text{max}}=70.25^\circ$, $Cu_{\text{K}\alpha}$ radiation, $\lambda=1.54178$ Å, $2\theta/\omega$ scans, T=296(2) K, absorption correction: none, 29 638 reflections collected, 5709 independent. Refinement on F^2 for 5709 reflections and 293 parameters gave GOF=0.996, R1=0.0334, wR2=0.0992 for $I>2\sigma(I)$. Residual electron density $-0.391<\Delta\rho<0.775$ e ų. Absolute structure parameter -0.008(5).
- [10] M. Brookhart, M. Wagner, J. Am. Chem. Soc. 1994, 116, 3641 3642.
- [11] Crystal structure for C₆₆H₄₇BF₂₄FeNPPdS {[(1a)Pd(Me)(NC-Ph)] $^+$ [(Ar F_4)B] $^-$ }: crystal size $0.23 \times 0.14 \times 0.10$ mm, monoclinic, space group $P2_1$, a = 17.372(3), b = 23.423(4), c = 18.140(3) Å, $\hat{\beta} = 110.615(3)^{\circ}$, $V = 6909(2) \text{ Å}^3$, Z = 4, $\rho_{\text{calcd}} = 1.487 \text{ Mg m}^{-3}$, $\mu = 10.615(3)^{\circ}$ 0.628 mm^{-1} , $2\theta_{\text{max}} = 56.56^{\circ}$, $Mo_{K\alpha}$ radiation, $\lambda = 0.71073 \text{ Å}$, $2\theta/$ ω scans, T=293 K, absorption correction: none, 24303 reflections collected, 10937 independent. Refinement on F^2 for 24303 reflections and 1739 parameters gave GOF = 1.002, R1 = 0.0895, wR2 = 0.2353 for $I > 2\sigma(I)$. Residual electron density -1.655 < $\Delta \rho < 0.944 \text{ e Å}^{-3}$. Absolute structure parameter -0.01(3). CCDC-231878 and CCDC-229828 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/conts/retrieving.html (or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB21EZ, UK; fax: (+44)1223-336-033; or deposit@ccdc.cam.ac.uk).
- [12] The absolute configuration of the alcohols **2–4** and **6a** was established by comparison with the data obtained by HPLC using a chiral stationary phase for both enantiomers of these compounds (see references [5c] and [5d]).
- [13] In all these reactions the enantioselectivity achieved from [(1e)Pd(Cl)(Me)] was significantly higher than that from the parent diphenylphosphanyl catalyst [(1a)Pd(Cl)(Me)].
- [14] A higher reactivity was found in DCE than in toluene, probably as a result of the increased solubility of the cationic Pd complex in the former.